

CHEMISTRY PRINCIPLES AND REACTIONS 7TH EDITION SOLUTIONS MANUAL pdf

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Unlike static PDF Chemistry: Principles And Reactions 7th Edition solution manuals or printed answer keys, our experts show you how to solve each problem step-by-step. No need to wait for office hours or assignments to be graded to find out where you took a wrong turn.

I doubt you would believe that. I could also tell you that the text is so clearly written, so attuned to the students in the twenty-first century that your students will learn chemistry with little or no effort on their part. You certainly would not believe that. I can tell you that the two goals in writing this edition have been to make it as clear and as interesting as possible. I hope you believe that, because it is true. Text messaging and Twitter™ have strongly influenced sentence length and structure. In current writing and conversation, short sentences or sentence fragments convey straight to-the-point information. Multimedia presentations are a way of life. Why Write a Short Book? The students and their parents justifiably do not want to pay for page books with material that is never discussed in the courses taught with those texts. The common perception is that a short book is a low-level book. I believe, however, that treating general concepts in a concise way can be done without sacrificing depth, rigor, or clarity. The criterion for including material continues to be its importance and relevance to the student, not its difficulty. To achieve this, the following guidelines are used. Eliminate repetition and duplication wherever possible. This simplifies not only the treatment of gaseous equilibrium but also the discussion of reaction spontaneity Chapter 16 and electrochemistry Chapter Relegate to the Appendices or Beyond the Classroom essays topics ordinarily covered in longer texts. Experience has shown and continues to show that although this approach is important to chemical bonding, most general chemistry students do not understand it but only memorize the principles discussed in the classroom. This material is of little value in a beginning course and is better left to a course in organic chemistry. This is summarized in a few pages in an essay in Chapter 15 in the Beyond the Classroom section. An extended discussion of the qualitative scheme and the chemistry behind it belongs in a laboratory manual, not a textbook. This material is traditionally covered in the last chapter of general chemistry texts. Although there are several biochemical topics included in the text among them a discussion of heme in Chapter 19 and carotenoids in Chapter 6, an entire chapter is not devoted to biochemistry. Interesting as this material is, it requires a background in organic chemistry that first-year students lack. Avoid superfluous asides, applications to the real world, or stories about scientists in the exposition of principles. There are many applications incorporated in the context of problems and some of the exposition of general principles. In general, however, a bare-bones approach is used. Students can easily be distracted by interesting but peripheral tidbits while they are striving hard to understand the core concepts. Favorite real-world applications and personal stories about scientists are in separate sections, Beyond the Classroom and Chemistry: They do admit to enjoying the marginal notes too.

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