

1: Petroleum - Wikipedia

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Tutorial Petroleum Coke Petrography Delayed coking is a thermal cracking process. The feed is generally petroleum derived. The carbonization reactions involve dehydrogenation, rearrangement and polymerization. Two of the common feedstocks are vacuum, residues and aromatic oils. The vacuum residues contain asphaltic compounds which are mostly heterocyclic molecules. The aromatic oils such as decant oil and pyrolysis residues have concentrations of polynuclear aromatics mainly with 6-carbon aromatic rings. They produce graphitic structures. In the delayed coker the feed enters the bottom of the fractionator where it mixes with recycle liquid condensed from the coke drum effluent. It is pumped through the coking heater then to one of two coke drums through a switch valve. Cracking and polymerization take place in the coke drum in a nominal hour period. Coking is a batch operation carried out in two coke drums. Coking takes place in one drum in 24 hours while decoking is carried out in the other drum. A complete cycle is 48 hours. Coke is cut from the drum using high pressure water. There are various amounts of sponge, needle and shot coke produced in delayed cokers. Virgin petroleum feedstocks have a large number of cross-linkages with less than 6 carbon atoms. These feedstocks tend to produce isotropic or amorphous cokes and when they are visibly very porous they are called sponge coke. The highly aromatic fractions from refinery cracking remain plastic longer during carbonization allowing crystals to form needle-like ocular structures. They have a low coefficient of thermal expansion CTE and maintain high current densities in products high conductivity. Shot coke is an abnormal type of coke resembling small balls. It is believed that residuum high in asphaltines and low API favor shot coke formation. Mesophase and its precursors have a wide size range depending on how rapid or slow components react and solidify. Fluidization in the coke drums may cause shot coke. It tends to be more isotropic, or fine granular in texture and is hard with a low porosity and high density which makes it difficult to crush. Coking severity decreases from bottom to top in the drum so bottom coke is more dense with lower VCM than top coke. Gas bubble percolation may account for some porous or spongy cokes. The feed vapors are cracked while forming a liquid film on the coke particles. The particles grow by layers until they are removed and new seed coke particles are added. Coke for the aluminum industry is calcined to less than 0.

2: petroleum-derived hydrocarbons - German translation â€“ Linguee

Petroleum is used mostly, by volume, for refining into fuel oil and gasoline, both important "primary energy" sources. 84 percent by volume of the hydrocarbons present in petroleum is converted into energy-rich fuels (petroleum-based fuels), including gasoline, diesel, jet, heating, and other fuel oils, and liquefied petroleum gas.

Oil derrick in Okemah, Oklahoma , Petroleum, in one form or another, has been used since ancient times, and is now important across society, including in economy, politics and technology. The rise in importance was due to the invention of the internal combustion engine , the rise in commercial aviation , and the importance of petroleum to industrial organic chemistry, particularly the synthesis of plastics, fertilisers, solvents, adhesives and pesticides. More than years ago, according to Herodotus and Diodorus Siculus , asphalt was used in the construction of the walls and towers of Babylon ; there were oil pits near Ardericca near Babylon , and a pitch spring on Zacynthus. Ancient Persian tablets indicate the medicinal and lighting uses of petroleum in the upper levels of their society. The use of petroleum in ancient China dates back to more than years ago. In I Ching , one of the earliest Chinese writings cites that oil in its raw state, without refining, was first discovered, extracted, and used in China in the first century BCE. In addition, the Chinese were the first to use petroleum as fuel as early as the fourth century BCE. The still active Erdpechquelle, a spring where petroleum appears mixed with water has been used since , notably for medical purposes. Oil sands have been mined since the 18th century. Unconventional reservoirs such as natural heavy oil and oil sands are included. Chemist James Young noticed a natural petroleum seepage in the Riddings colliery at Alfreton , Derbyshire from which he distilled a light thin oil suitable for use as lamp oil, at the same time obtaining a more viscous oil suitable for lubricating machinery. In Young set up a small business refining the crude oil. Young found that by slow distillation he could obtain a number of useful liquids from it, one of which he named "paraffine oil" because at low temperatures it congealed into a substance resembling paraffin wax. Romania is the first country in the world to have had its annual crude oil output officially recorded in international statistics: Advances in drilling continued into when local driller Shaw reached a depth of 62 metres using the spring-pole drilling method. Access to oil was and still is a major factor in several military conflicts of the twentieth century, including World War II , during which oil facilities were a major strategic asset and were extensively bombed. Petroleum also makes up 40 percent of total energy consumption in the United States, but is responsible for only 1 percent of electricity generation. Viability of the oil commodity is controlled by several key parameters, number of vehicles in the world competing for fuel, quantity of oil exported to the world market Export Land Model , net energy gain economically useful energy provided minus energy consumed , political stability of oil exporting nations and ability to defend oil supply lines. While significant volumes of oil are extracted from oil sands, particularly in Canada, logistical and technical hurdles remain, as oil extraction requires large amounts of heat and water, making its net energy content quite low relative to conventional crude oil. Under surface pressure and temperature conditions , lighter hydrocarbons methane , ethane , propane and butane exist as gases, while pentane and heavier hydrocarbons are in the form of liquids or solids. However, in an underground oil reservoir the proportions of gas, liquid, and solid depend on subsurface conditions and on the phase diagram of the petroleum mixture. Because the pressure is lower at the surface than underground, some of the gas will come out of solution and be recovered or burned as associated gas or solution gas. A gas well produces predominantly natural gas. However, because the underground temperature and pressure are higher than at the surface, the gas may contain heavier hydrocarbons such as pentane, hexane , and heptane in the gaseous state. At surface conditions these will condense out of the gas to form " natural gas condensate ", often shortened to condensate. Condensate resembles gasoline in appearance and is similar in composition to some volatile light crude oils. Many oil reservoirs contain live bacteria.

3: Petroleum naphtha - Wikipedia

Microporous carbons were prepared from petroleum coke by carbonization and activation. The synthesized microporous carbon showed high CF 4 adsorption uptake. Fast adsorption-desorption kinetics and excellent cyclic stability were achieved.

Petrolatum, petroleum jelly Petrolatum, petroleum jelly Petrolatum, or petroleum jelly, derived from petroleum, is often used in personal care products as a moisturizing agent. When properly refined, petrolatum has no known health concerns. However, petrolatum is often not fully refined in the US, which means it can be contaminated with toxic chemicals called polycyclic aromatic hydrocarbons PAHs. Petrolatum is a byproduct of petroleum refining. These qualities make petrolatum a popular ingredient in skincare products and cosmetics. However, with an incomplete refining history, petrolatum could potentially be contaminated with polycyclic aromatic hydrocarbons, or PAHs. PAHs are byproducts of organic material combustion, commonly stored in fats upon exposure due to its lipophilic properties. The primary concern with petrolatum is the potential contamination with PAHs. The EU mandates that for cosmetic use, the full refining history of the petrolatum must be known and proven to be non-carcinogenic. The US sets no requirements on refinement and the PAH content in the petrolatum used in personal care products. Avoid products with petrolatum, unless the company clearly indicates petrolatum is fully refined as white petrolatum on the label or their company website. References [1] Japour, M. Petroleum Refining and Manufacturing Processes pp. Wetzel Publishing Company, Incorporated. The science of black hair: A comprehensive guide to textured hair pp. Report on Carcinogens, Thirteenth Edition. Research Triangle Park, NC: Agents classified by the IARC monographs, volumes 1â€” Retrieved August 10,

4: Petroleum Coke Tutorial | Petrographic Atlas | SIU

*Petroleum Derived Carbons (Acs Symposium Series) by John D. Bacha. Amer Chemical Society, *Price HAS BEEN reduced by 10% until Monday, Nov. 5 sale item* pp., hardcover, ex library, else text clean & amp; binding tight.*

Received Jan 23; Accepted Mar Abstract Petroleum coke is a valuable and potential source for clean energy storage if it could be modified legitimately and facilely. In the present study, porous carbon with high surface area and abundant oxygen-containing groups was prepared from petroleum coke by chemical activation and modification processes. It presents a high specific capacitance and excellent rate performance in KOH electrolyte. In addition, the energy density of this material in aqueous electrolyte can be as high as The high energy density and excellent rate performance ensure its prosperous application in high-power energy storage system. Petroleum coke, Activated carbon, Oxygen doping, Supercapacitor, Pseudo-capacitance Background Environment-friendly energy supply is seemed to be one of the biggest concerns right now, which is closely associated with our lives [1]. It is undeniable that fossil energy will continue to play a major role in meeting our energy requirements for an extended period. In addition, portable electronic devices and hybrid electric vehicles, with power source of electric energy resource, are growing very fast in recent years. In this regard, efficient electrical energy storage systems have gradually caused extensive concern [1 , 3 â€” 5]. They can play a vital role in some applications that needed high-power delivery while batteries cannot meet the requirement. Supercapacitors can be divided into two categories according to the charge storage mechanism [8]. One is electrical double-layer capacitors EDLCs which store energy by the adsorption of both anions and cations. The other category is the pseudo-capacitors that store energy through fast surface redox reactions. Carbon-based materials have been widely investigated as electrodes of supercapacitors due to their desirable physical and chemical properties [9 â€” 12]. These properties include low cost, ease in processing, controllable porosity, and electrocatalytic active sites for a variety of redox reactions. Large accessible specific surface area and appropriate pore size of the carbon-based electrodes are crucial to ensure a good performance of EDLCs in terms of both power delivery rate and energy storage capacity. Activated carbons ACs are often considered as EDLC electrode materials because of their high surface area and somewhat controllable pore size. For example, a mesoporous activated carbon sphere derived from resorcinol-formaldehyde resin has been prepared and chosen as electrode of EDLC [13]. It possesses a high surface area and presents a good electrochemical double-layer capacitive performance. Carbon precursors like such resin are too numerous to mention [14 â€” 21]. Among them, petroleum coke PC is a better candidate for preparing high surface area AC due to its high carbon content and low ash content. Lee and Choi [20] prepared a high surface area activated carbon derived from high-sulfur petroleum cokes by chemical activation using KOH. It is obvious that PC is very difficult to be activated due to its stable micrographic structure and lack of the initial pores [22]. During the activation of PC, large quantity of KOH was expended to obtain high surface area AC, which was uneconomic and eco-unfriendly. In addition, pure electrochemical double layer EDL capacitance from AC is relatively low, resulting in an inferior energy density. Heteroatom doping seems to be an effective way to solve these problems. It can not only make PC easy to be activated [23] but also introduce pseudo-capacitance to enhance the overall capacitance of the electrode materials. The as-prepared AC possesses a high level of nitrogen and large surface area, leading to an improved CO₂ uptake capacity. The resulting activated carbon had higher specific surface area and better iodine adsorption value. Unfortunately, the influence of functionalization of PC-derived carbon on its energy storage performance has rarely been studied, limiting its wide use in clean energy storage applications. Therefore, the purpose of this research is to investigate the effects of oxygen doping of petroleum coke-derived carbon PCAC on its capacitive performance. The introduced functional groups not only enhance the EDL capacitance but also provide pseudo-capacitance, leading to an enhanced overall capacitance. Synthesis of Surface-modified Activated Carbon Derived from PC The samples were prepared from PC with a combination of KOH activation and chemical modification using hydrogen peroxide or nitric acid as oxidizing agent. The carbonized sample was washed and filtered with HCl aqueous solution and distilled water successively until the pH of the filtrate was

7. Chemical Modification In a typical procedure, 3. The surface chemical properties of the samples were characterized by Fourier transform infrared spectroscopy FT-IR Nicolet , Thermoscientific. Nitrogen adsorption-desorption measurements were performed on a Tristar analyzer Micromeritics, USA to obtain specific surface area and pore structure parameters of the samples. Electrochemical Measurement The electrochemical capacitive performance measurements were carried out by a CHI D instrument using a three-electrode system. The three-electrode system includes platinum film as the counter electrode, saturated calomel electrode SCE as the reference electrode, and the as-prepared sample as the working electrode. The carbon materials were mixed with polytetrafluoroethylene as a binder at a weight ratio of SCE was selected for the electrochemical measurements. The average specific capacitance C was derived from the discharge curve according to the following equation:

5: Petroleum Hydrocarbons (PHCs) | Chemistry Matters

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Physical and Chemical Properties of Petroleum I. Hydrocarbons- substance made of hydrogen and carbon among other elements Grade through three states of matter: Petroleum exploration largely concerned with the fluids gases and liquids. Several specific forms of hydrocarbons- Dry gas- contains largely methane, specifically contains less than 0. Wet gas- contains ethane propane, butane. Up to the molecular weight where the fluids are always condensed to liquids Condensates- Hydrocarbon with a molecular weight such that they are gas in the subsurface where temperatures are high, but condense to liquid when reach cooler, surface temperatures. Liquid hydrocarbons- commonly known as oil, or crude oil, to distinguish it from refined hydrocarbon products. Plastic hydrocarbons- asphalt Solid hydrocarbons- coal and kerogen- kerogen strictly defined is disseminated organic matter in sediments that is insoluble in normal petroleum solvents. Gas hydrates- Solids composed of water molecules surrounding gas molecules, usually methane, but also H₂S, CO₂, and other less common gases. Given a strict definition by the petroleum industry- "a mixture of hydrocarbons and varying quantities of nonhydrocarbons that exists either in the gaseous phase or in solution with crude oil in natural underground reservoirs". The common gasses in reservoirs can be divided based on their origins: Dissolved gas- That portion of natural gas that is dissolved in liquid phase in the sub- surface. It can be and usually is physically separated from the liquid when the fluids are produced. Associated gas- Also known as the "gas cap" is free gas not dissolved that sits on top of, and in contact with, crude oil in the reservoir. Non-associated gas- Free gas that is trapped without a significant amount of crude oil. Natural gas liquids- The liquids that can be, and are liquified, in the field and at gas processing plants. Include the wet gases, natural gasoline, and condensate. Generalities of the origin of hydrocarbons: Inorganic and organic 1 Inorganic- Hydrocarbons form from reduction of primordial carbon or oxidized forms at high temperatures in the earth 2 Organic- accumulation of hydrocarbons produced directly by living organisms, as well as the thermal alteration of biologically formed organic matter. It is generally recognized that most hydrocarbons are produced by the organic method. A few hydrocarbons in the crust may be from inorganic sources, but the majority of them are from organic. See Handout Hunt, Fig. They generally contain more than 15 C atoms, and are easily recognized structures biomarkers. The thermal alteration of formed hydrocarbons continues with continued burial depth. Natural Gases- Their components A. Hydrocarbon gases Hydrocarbon gases are largely composed of the paraffin series- straight and branched, single bonded chains of hydrocarbons. Look at Table 2. It can form in three ways: Derived from the mantle presumably primordial methane. Commonly assumed to form by the other two processes: Large caused by the reduction of CO₂ during oxidation of the organic matter. Appears that the thermal degradation has to be catalyzed for it to occur in nature. Microbial methane can be distinguished from thermogenic methane on the basis of its isotopic composition. Isotopic composition controlled by various fractionation processes- Microbes create large fractionation- there is a vital effect and the reaction occurs at lower temperature with larger fractionation factor Thermogenic breakdown has little fractionation- breakdown occurs at high temperature and there is no vital effect. Rarely perhaps never formed by bacterial processes. Thus the presence of heavier hydrocarbons in natural gas probably reflects proximity to liquid hydrocarbon reservoir. Gases are important during drilling of wells- 1 commonly overpressured, can cause blowouts 2 Useful way to identify producing or hydrocarbon bearing horizons. Extract gas from drilling mud, run through chromatograph, identify the amount and type of gases entering the well bore. Non-hydrocarbon gases 1 Noble Gases- Helium, Argon, and Radon These gases are inert- do not take part in chemical reactions. They originate from decay of radioactive isotopes of various elements, predominately the U series elements. Not all steps shown, only those with alpha decay: These elements also have similar decay schemes and will produce He for each alpha decay. Note that U is part of the U series beta decay, and not listed as a separate decay series. Each step of the three decay schemes produce one He atom.

May be surprising that there is not more He present in subsurface reservoir. The He atom is light, energetic and is difficult to trap in the subsurface. He is economic when in high enough concentrations. Produced from the Hugoton field in the Panhandle of Texas. Argon is produced by the radioactive decay of K beta decay. Very big scientific significance K-Ar dating, e. Radon is part of U decay scheme. Also no economic significance, but might be a major health problem. The oxidation-reduction reaction includes Ferric Fe redbeds- Nitrogen-bearing deposits are commonly associated with redbeds. There is no economic significance to the nitrogen other than it reduces the concentrations of economic hydrocarbons. It must be actively produced within in reservoir, adjacent source beds, or diffusing upward from depth. Largely from the decomposition of oxygen bearing groups in organic matter. Usually derived from continentally derived organic matter. CO₂ now used for enhanced oil recovery so it can be economic. Combination of H₂S, CO₂, and water easily corrode metal. Also must be disposed of safely. Reduces the value of the hydrocarbon deposit

Origin: One way to write such a reaction: It can also be converted to sulfur metal, which is sold. Gas hydrates

Compounds of frozen water and gas. Called a "clathrate" structure. Two structures of hydrates: Now larger void space accommodates larger gases up through pentanes and n-butanes. Thermodynamic stability of hydrates are what make them important: They can be observed in seismic sections as a BSR "Bottom simulating reflector". BSR are generally believed to be the transition from solid hydrate to free gas at the point in the sediment where the hydrate dissociates. Need to assume composition of water and gas. Possibility that hydrates are the single largest reservoir of methane on the earth. Thus contain more molecules of methane in small area than the gas form of methane. Require techniques for removing hydrate from the sediments- change their stability Also b they greatly reduce the permeability of the sediment. May act as a seal for reservoirs.

Crude Oil 1 Appearance- color- yellow, green, brown to black. In general, viscosity and API gravity are inversely related. Other components include sulfur, oxygen, hydrogen and other elements. The limited chemical composition is misleading- there are hundreds of different compounds that can be generated from C and H. These compounds divided into 1 Hydrocarbons, which contain only hydrogen and carbon and 2 Heterocompounds, which contain elements in addition to H and C. Largest molecule recorded from crude oil contains 78 carbons. Two types of alkane isomers ie molecules with identical compositions, but different structures: Straight chain, called "normal alkanes" e. They are saturated as well- only single bonds between carbons. They have a sweet smell- thus named aromatics. They can be modified by substituting a alkane for one of the hydrogens. Oxygen can range between 0. These elements are not other compounds e. H₂S or free N or contaminants There are also some metals, but only Nickel and Vanadium have been shown to be part of the compounds and not contaminants.

Get this from a library! Petroleum derived carbons: a symposium co-sponsored by the Division of Petroleum Chemistry, inc. and the Division of Industrial and Engineering Chemistry at the th meeting of the American Chemical Society, Philadelphia, Penn., April ,

This mixture of chemicals can also be described by common chemical characteristics such as boiling point ranges or size of the molecules. These characteristics are very important for determining the value of a crude oil. For example, a light crude contains mainly small, valuable chemicals, whereas a heavy bitumen contains mainly larger, less valuable chemicals, which must be refined more thoroughly to add value. The chemical characteristics of PHCs are also important when assessing potential impacts of spilled material. However, this poses a problem, as it is simply not practical to measure each and every chemical associated with PHCs. Certain groups of chemicals within PHCs present specific toxicological hazards, for example benzene and toluene, and the polycyclic aromatic hydrocarbons PAHs , and these chemicals are individually measured in a PHC mixture. The remaining PHCs are often grouped together with a specific health guideline for the group or groups. The nature of how this is carried out is not globally unified, and the details can be rather complicated. Measurement methods vary between countries, between provinces and states within a country, and even tailored for a specific location, depending on the nature of the PHCs. Each fraction has a specific guideline for toxicological hazards, so that overall the method can account for the different hazards associated with, for example, a light crude oil and a heavy bitumen oil. In the US, TPHs are often separated into specific classes such as gasoline range organics or diesel range organics. As with the Canadian method, this attempts to account for the different hazards associated with different types of PHC product. However, some US states require the separation of different classes of chemicals called aliphatics and aromatics. PHCs also act like a fingerprint of the oil product and are commonly used in environmental forensics investigations. Specific groups of chemicals can be used to provide a forensic determination of where a spill may have originated, the age of the spill and ultimately linking the spill to a responsible party. The nature of the living material that originally died and decayed, millennia ago, The geological mechanisms of heat, pressure and chemical catalysis that converted the organic material to produce oil, Upgrading and refining of oil, Environmental weathering, such as evaporation and microbiological alteration. Different types of hydrocarbon sources can be identified by examining the general composition of the PHCs. At its simplest level this can be done by analyzing the GC chromatogram, or extracted ion chromatograms and comparing this to examples of different product types. We have used this type of analysis to distinguish many different types of impacts, such as gasoline and diesel components of a mixed impacted site as part of a source allocation project. A more thorough and more precise analysis can be completed when analyzing specific groups of chemicals within the PHCs. Many of these are called biomarkers. These are chemicals that retain some general structure from the biology. Comprehensive analysis of biomarkers presents a key method of determining the origin of a spill or to characterize oil products. Other important chemical groups such as PAHs and naphthenic acids can also help distinguish very similar oil products. With the overwhelming number of potential unique markers present in oil and refined products, in depth knowledge of these components allows the trained environmental forensics expert to distinguish between even very similar sources.

7: Petrolatum, petroleum jelly - Safe Cosmetics

Petroleum coke is a valuable and potential source for clean energy storage if it could be modified legitimately and facilely. In the present study, porous carbon with high surface area and abundant oxygen-containing groups was prepared from petroleum coke by chemical activation and modification processes.

Petroleum crude oils are complex mixtures of hydrocarbons, chemical compounds composed only of carbon C and hydrogen H. Nineteenth-century chemists classified hydrocarbons as either aliphatic or aromatic on the basis of their sources and properties. Aromatic hydrocarbons constituted a group of related substances obtained by chemical degradation of certain pleasant-smelling plant extracts. The terms aliphatic and aromatic are retained in modern terminology, but the compounds they describe are distinguished on the basis of structure rather than origin. Aliphatic hydrocarbons are divided into three main groups according to the types of bonds they contain: Alkanes have only single bonds, alkenes contain a carbon-carbon double bond, and alkynes contain a carbon-carbon triple bond. Aromatic hydrocarbons are those that are significantly more stable than their Lewis structures would suggest; i. This classification of hydrocarbons serves as an aid in associating structural features with properties but does not require that a particular substance be assigned to a single class. Indeed, it is common for a molecule to incorporate structural units characteristic of two or more hydrocarbon families. A molecule that contains both a carbon-carbon triple bond and a benzene ring, for example, would exhibit some properties that are characteristic of alkynes and others that are characteristic of arenes. Alkanes are described as saturated hydrocarbons, while alkenes, alkynes, and aromatic hydrocarbons are said to be unsaturated. In order of increasing number of carbon atoms, methane CH_4 , ethane C_2H_6 , and propane C_3H_8 are the first three members of the series. Methane, ethane, and propane are the only alkanes uniquely defined by their molecular formula. For C_4H_{10} two different alkanes satisfy the rules of chemical bonding namely, that carbon has four bonds and hydrogen has one in neutral molecules. One compound, called n-butane, where the prefix n- represents normal, has its four carbon atoms bonded in a continuous chain. The other, called isobutane, has a branched chain. Different compounds that have the same molecular formula are called isomers. Isomers that differ in the order in which the atoms are connected are said to have different constitutions and are referred to as constitutional isomers. An older name is structural isomers. The compounds n-butane and isobutane are constitutional isomers and are the only ones possible for the formula C_4H_{10} . Because isomers are different compounds, they can have different physical and chemical properties. There is no simple arithmetic relationship between the number of carbon atoms in a formula and the number of isomers. The number of constitutional isomers increases sharply as the number of carbon atoms increases. There is probably no upper limit to the number of carbon atoms possible in hydrocarbons. The alkane $\text{CH}_3(\text{CH}_2)_n\text{CH}_3$, in which carbon atoms are bonded in a continuous chain, has been synthesized as an example of a so-called superlong alkane. Several thousand carbon atoms are joined together in molecules of hydrocarbon polymers such as polyethylene, polypropylene, and polystyrene. Number of possible alkane isomers molecular formula.

8: - Petroleum Derived Carbons (Acs Symposium Series) by John D. Bacha

However, the contribution of thermally matured petroleum-derived OC to the sedimentary OC pool is also evident, especially in the southern part of the study area as shown by the low carbon preference index (CPI, <1) and natural n-alkanes ratio (NAR, <) values.

Java mini projects with source code Shifting sands of history Recycling of waste material in india The Parents Guide to Speech and Language Problems Concluding note: neither weight nor weight loss. A history of the class that graduated at Princeton Theological Seminary, in the year 1864 Entrepreneurs road to business success and personal freedom Kerala model of development Microbiology book by cp baveja Microsoft flight simulator manual An account of oneself. Scenes of address Tarpon Fishing in Mexico and Florida Gender identities and womens agency in early modern tithe dispute. Whirlpool dishwasher tech sheet w10461429a Daily practice spoken english It project management life cycle Selection training of the interviewer How to teach your baby to read The Ritual Of The Ladies Auxiliaries To The Local Aeries Of The Fraternal Order Of Eagles Extracteur page dun Making it work by connecting parents Only love is real : Plato, Kant, and The matrix trilogy James Lawler Clinical companion for Fundamentals of nursing Estate of Moses S. McCord, deceased. Period contemporary patterns for fashion dolls Guilding, gaming, and girls Genesis Downey Early homes of New England Bad Connection (Secret Life Samantha McGregor) Digital Control of Electrical Drives (Power Electronics and Power Systems) Essentials of plastic surgery 2nd edition Miss Pickerell takes the bull by the horns Spruce beetle epidemiology and management in NW Alberta Frontiers in medical ethics Why Go Lisa Tucker A World in Focus The World Through Words Central South America (A World in Focus) Datsun owners workshop manual The hostiles were apparently everywhere Day the Revolution began Conditions of listening Fossils and Geologic Dating