

1: Polymer - Wikipedia

Polymers range from familiar synthetic plastics such as polystyrene to natural biopolymers such as DNA and proteins that are fundamental to biological structure and function. Polymers, both natural and synthetic, are created via polymerization of many small molecules, known as monomers.

Can you see through these real-life optical illusions? Polymers encompass a broad category of large molecules created by the bonding of many similar constituent molecules. There are many different kinds of polymer materials, including cellulose, natural or biopolymers, silicones and plastics. Polymer materials are often created or harvested for a specific purpose depending on the type. Many polymer materials are natural, existing in biological entities and in use for many years. Others are synthetic, created through chemical or industrial processes to perform certain functions. Cellulose is typically found in wood and plants as a natural part of those materials. Once derived, it can be used to create products such as fiber board, paper and cellophane. Polymer wood, for example, is typically created using cellulose and similar materials to achieve a specified strength or consistency. Polymers containing cellulose also can be considered dietary fiber when consumed by humans and may provide an important food for creatures such as termites. Other natural polymers have been in use for a long time. Examples include shellac, rubber and amber. Shellac is usually derived from the secretions of a specific bug and is often used as a wood finish. Rubber is most often derived from naturally occurring latex and can be further refined to make many products. Amber is generally formed from fossilized tree resin and has been used for many years as decoration, medicine and jewelry. Ad Much like plant- or bug-based polymer materials, nucleic acids and proteins also are considered polymers. More specifically, they are considered biopolymers, because they are produced by organisms. Like other polymer materials, nucleic acids consist of repeated molecular chains. The combinations of these acids typically become the biopolymer proteins used to build most organisms. Silicones, on the other hand, are considered to be synthetic polymer materials that typically consist of silicon combined with carbon , oxygen or hydrogen to create new materials. These polymers have many uses, depending on the molecular chain configuration, but those uses may include toys, plumbing materials and lubricants. There are many polymer materials based on the combination of silicon with other elements. Plastic materials are polymers that may be organically based or almost entirely synthetic. Polymer materials considered to be plastic are generally considered important in modern life, and many materials and products would not exist without plastic materials. Examples of these polymers can include polyvinyl chloride PVC , nylon and polystyrene. Most plastic polymer materials have a high molecular mass, which gives related products rigidity and plasticity.

2: Polymer | Definition of Polymer by Merriam-Webster

A polymer is a large molecule that is made up of repeating subunits connected to each other by chemical bonds. Do you need some examples of polymers? Here is a list of materials that are polymers, plus some examples of materials that are not polymers.

Polymer classes Polymers are of two types: Natural polymeric materials such as shellac , amber , wool , silk and natural rubber have been used for centuries. A variety of other natural polymers exist, such as cellulose , which is the main constituent of wood and paper. The list of synthetic polymers , roughly in order of worldwide demand, includes polyethylene , polypropylene , polystyrene , polyvinyl chloride , synthetic rubber , phenol formaldehyde resin or Bakelite , neoprene , nylon , polyacrylonitrile , PVB , silicone , and many more. More than million tons of these polymers are made every year However, other structures do exist; for example, elements such as silicon form familiar materials such as silicones, examples being Silly Putty and waterproof plumbing sealant. Oxygen is also commonly present in polymer backbones, such as those of polyethylene glycol , polysaccharides in glycosidic bonds , and DNA in phosphodiester bonds. Polymerization The repeating unit of the polymer polypropylene Polymerization is the process of combining many small molecules known as monomers into a covalently bonded chain or network. During the polymerization process, some chemical groups may be lost from each monomer. This is the case, for example, in the polymerization of PET polyester. The distinct piece of each monomer that is incorporated into the polymer is known as a repeat unit or monomer residue. Laboratory synthetic methods are generally divided into two categories, step-growth polymerization and chain-growth polymerization. However, some newer methods such as plasma polymerization do not fit neatly into either category. Synthetic polymerization reactions may be carried out with or without a catalyst. Laboratory synthesis of biopolymers, especially of proteins , is an area of intensive research. Biopolymer Microstructure of part of a DNA double helix biopolymer There are three main classes of biopolymers: In living cells, they may be synthesized by enzyme-mediated processes, such as the formation of DNA catalyzed by DNA polymerase. The synthesis of proteins involves multiple enzyme-mediated processes to transcribe genetic information from the DNA to RNA and subsequently translate that information to synthesize the specified protein from amino acids. The protein may be modified further following translation in order to provide appropriate structure and functioning. There are other biopolymers such as rubber , suberin , melanin and lignin. Modification of natural polymers[edit] Naturally occurring polymers such as cotton, starch and rubber were familiar materials for years before synthetic polymers such as polyethene and perspex appeared on the market. Many commercially important polymers are synthesized by chemical modification of naturally occurring polymers. Prominent examples include the reaction of nitric acid and cellulose to form nitrocellulose and the formation of vulcanized rubber by heating natural rubber in the presence of sulfur. Ways in which polymers can be modified include oxidation , cross-linking and endcapping. Especially in the production of polymers the gas separation by membranes has acquired increasing importance in the petrochemical industry and is now a relatively well-established unit operation. The process of polymer degassing is necessary to suit polymer for extrusion and pelletizing, increasing safety, environmental, and product quality aspects. Nitrogen is generally used for this purpose, resulting in a vent gas primarily composed of monomers and nitrogen. A second set of properties, known as microstructure , essentially describes the arrangement of these monomers within the polymer at the scale of a single chain. These basic structural properties play a major role in determining bulk physical properties of the polymer, which describe how the polymer behaves as a continuous macroscopic material. Chemical properties, at the nano-scale, describe how the chains interact through various physical forces. At the macro-scale, they describe how the bulk polymer interacts with other chemicals and solvents. Monomers and repeat units[edit] The identity of the repeat units monomer residues, also known as "mers" comprising a polymer is its first and most important attribute. Polymer nomenclature is generally based upon the type of monomer residues comprising the polymer. Polymers that contain only a single type of repeat unit are known as homopolymers, while polymers containing two or more types of repeat units are known as copolymers. Ethylene-vinyl acetate , on the other

hand, contains more than one variety of repeat unit and is thus a copolymer. Some biological polymers are composed of a variety of different but structurally related monomer residues; for example, polynucleotides such as DNA are composed of four types of nucleotide subunits. A polymer molecule containing ionizable subunits is known as a polyelectrolyte or ionomer.

Microstructure The microstructure of a polymer sometimes called configuration relates to the physical arrangement of monomer residues along the backbone of the chain. Structure has a strong influence on the other properties of a polymer. For example, two samples of natural rubber may exhibit different durability, even though their molecules comprise the same monomers.

Polymer architecture Branch point in a polymer An important microstructural feature of a polymer is its architecture and shape, which relates to the way branch points lead to a deviation from a simple linear chain. Types of branched polymers include star polymers, comb polymers, brush polymers, dendronized polymers, ladder polymers, and dendrimers. A variety of techniques may be employed for the synthesis of a polymeric material with a range of architectures, for example Living polymerization.

Chain length[edit] The physical properties [24] of a polymer are strongly dependent on the size or length of the polymer chain. Since synthetic polymerization techniques typically yield a polymer product including a range of molecular weights, the weight is often expressed statistically to describe the distribution of chain lengths present in the same. Common examples are the number average molecular weight and weight average molecular weight.

Monomer arrangement in copolymers[edit] Main article: A copolymer containing a controlled arrangement of monomers is called a sequence-controlled polymer. Alternating copolymers possess two regularly alternating monomer residues: An example is the equimolar copolymer of styrene and maleic anhydride formed by free-radical chain-growth polymerization. A statistical copolymer in which the probability of finding a particular type of monomer residue at a particular point in the chain is independent of the types of surrounding monomer residue may be referred to as a truly random copolymer [38] [39] structure 3. For example, the chain-growth copolymer of vinyl chloride and vinyl acetate is random. Polymers with two or three blocks of two distinct chemical species e. Polymers with three blocks, each of a different chemical species e. Graft or grafted copolymers contain side chains or branches whose repeat units have a different composition or configuration than the main chain.

Tacticity Tacticity describes the relative stereochemistry of chiral centers in neighboring structural units within a macromolecule. There are three types of tacticity:

3: Materials Science and Engineering: Polymers | Materials Science and Engineering

BU Polymer Materials focuses on supplying the rubber industry, cable producers, the PVC industry and friction material end users. We represent the leading manufacturers of high-quality materials, who are able to provide strong technical support for our customers.

Brake linings Aramids, acrylics Synthetic fiber volumes have grown at the expense of natural fibers. The drivers are lower costs and technical improvements, which allow the synthetics to emulate desirable natural fiber aesthetics while exhibiting superior in-use performance. The commodity markets are divided primarily among nylon, polyester, and polyolefin, with polyester emerging as the largest. Cost-performance and environmental considerations have led to a diminution in the use of cellulose and acrylics. The introduction of a new commodity fiber is generally regarded as unlikely. This same time period has seen the rapid growth of high-performance fiber technologies. These technologies fall into three classes: High-modulus, high-strength fibers based on rodlike, liquid crystalline nematogenic polymers. The most common examples are the lyotropic aramids and the thermotropic polyesters. These fibers are characterized by tensile moduli greater than 70 gigapascals GPa , tensile strengths on the order of 3 to 4 GPa, and low properties in compression or shear. Morphological manipulation of conventional polymers, such as high-molecular-weight

Page 80 Share Cite Suggested Citation: Polymer Science and Engineering: The Shifting Research Frontiers. The National Academies Press. Polymeric precursor fibers that can be converted to other chemical forms after spinning. The most common examples are acrylic fibers that can be converted to carbon fibers and a variety of silicon-containing polymeric fibers that can be converted to silicon carbide or silicon nitride fibers. Typical applications of high-performance fibers are composite reinforcement, ropes and cables, and antiballistic clothing. As a group, these fibers represent successful technical developments, but they have proved less commercially attractive than once believed for a variety of reasons. The spinning process can be described as follows. A polymer is first converted to a liquid through melting or dissolution, and the liquid is then continuously forced through a spinnerette a plate with many of small holes to form filaments. Most polymeric fibers are semicrystalline. If the polymer forms a stable melt, the process is called melt spinning. For polymers that degrade prior to melting, the polymer is spun from a solution; if the solvent is evaporated, the process is termed dry spinning; if the solution is coagulated in a nonsolvent bath, the process is termed wet spinning. Removal of the spinnerette from the wet spinning coagulation bath is the innovation known as dry-jet wet spinning. The ratio of final filament velocity to the initial filament velocity is termed the drawdown ratio. The principal parameters controlling the as-spun structure and, hence, properties of the as-spun filament are the rate of cooling and the applied stress. Crystallinity once formed can be further oriented by stretching and perfected through annealing. Key structural elements are the amount and orientation of crystalline regions, the orientation of noncrystalline regions, and connectivity between regions, tie molecules, and so on. Careful control of the sequence in which chains are oriented and crystallized has a profound effect on the microstructure produced. Such controlled processing allows, for example, the decoupling of crystalline and noncrystalline orientation, enabling fibers with high tensile modulus correlated with high crystalline orientation and low thermal shrinkage correlated with low noncrystalline orientation to be produced. Typical spinning speeds are thousands of meters per minute, typical melt drawdowns are on the order of , and typical solid-state draw ratios range from about 2 to 6 in conventional processing to greater than 50 in the production of certain high-performance products. High-performance fiber processing is characterized by maximizing axial chain orientation and minimizing

Page 81 Share Cite Suggested Citation: To control friction and static behavior in subsequent processing, a variety of oils or other surface treatments are applied to the fibers prior to take-up. The many complex processing steps of fibers add to the stress-temperature history of the fiber and hence significantly modify the end-use properties of the material. To a large extent, the conditions employed in spinning, in addition to the particular chemistry of the polymer being spun, determine the end-use performance of a fiber. Work on future fibers will focus on producing cost-performance improvements and product variants through morphological control rather than new chemistries. With the huge lengths of fibers

produced, process robustness and property uniformity have always been major issues; future products will make more use of advanced computerized process control and will operate in areas of property response that are less sensitive to minor process variation. Elimination of downstream process steps will lead to additional cost-performance improvements, for example, on-line texturing and surface modifications to meet specific friction or adhesion requirements. Environmental considerations will influence future fiber developments in a number of areas. The elimination of solvent-based processing will be driven by stricter emissions standards, as will the elimination of heavy metal catalysis. Novel processes based on very fast melting techniques e. The reduction of off-specification production will become more important as the cost of waste disposal increases and as easy-to-reclaim fibers grow in importance e. The future of high-performance fibers lies in the reduction of costs and the improvement of utilization. The former is best influenced by lower-cost monomers, and the latter through the development of manufacturing technologies that allow cost-effective part production from fiber-reinforced composites. High-performance fiber development will cease to be solely performance driven and will, as in the case of all other fibers, become driven by cost and performance. Silks, produced by worms and spiders, have attracted attention because they possess tensile properties similar to those of high-performance synthetic fibers but with much higher toughness. The use of recombinant DNA techniques allows silks of specific molecular architectures to be produced and their performance to be correlated with specific chemical and physical features. The increased structure-property insights gained from these studies should allow the definition of biomimetic fibers, based on other than naturally occurring amino acids, with greatly improved performance characteristics. Page 82 Share Cite Suggested Citation: Adhesives have been used for most of recorded history. They are mentioned in Egyptian hieroglyphics, in the Bible, and in the writings of the early natural philosophers. The physical strength of an assembly made by the use of adhesives, known as an adhesive joint, is due partly to the forces of adhesion, but primarily to the cohesive strength of the polymeric materials used to formulate the adhesive. Thus, the range of strengths available in adhesive joints is limited to the strengths of the polymers useful in the formulation of adhesives. Indeed, the technology of adhesives tracks well with the technology of polymers. As new polymers were synthesized, new adhesives were developed that used those polymers.

4: What Are the Different Types of Polymer Materials?

Materials as a field is most commonly represented by ceramics, metals, and polymers. While noted improvements have taken place in the area of ceramics and metals, it is the field of polymers that has experienced an explosion in progress. Polymers have gone from being cheap substitutes for natural.

These are excellent pages. You will need to have the plug-in Chemscape Chime installed on your browser to be able to see the macro-molecular views of polymers. Engineers the world over use heat-shrinkable tubing instead of standard approaches to insulation, such as taping or molding in place. The tubing comes in a wide range of sizes, colors, and materials. When heated, it shrinks to conform to the size and shape of the underlying material, making installation fast and easy. How much shrinkage can occur? Check out this Youtube video. Why do you think this plastic tubing has the ability to shrink? If you think you know the answer, check at the following website. Do you know that recent technological developments have lead to electrically conductive polymers? Semiconductor behavior is now possible using polymeric systems. For example, semiconducting polymers, sandwiched between two electrodes, can generate light of any color. The figure at the right shows polymer materials and their solutions top , a glass plate covered with a thin polymer film bottom and three operating displays of two different colours in the middle. This technology will lead to OLED organic light-emitting diode , flat panel displays. Such a display would be light in weight, less power consuming than other alternatives, and perhaps flexible. For more information about engineered polymers, visit the web pages of Philips Research. Can you think of applications for a OLED display? Maybe you will be the materials engineer on the team that figures out an economical and reliable means to produce a reasonably priced, three-pixel patterned display, using polymer technology and manufacturing knowhow. Polymers are materials comprised of long molecular chains. Most polymers are carbon based and have relatively low melting points. Polymers have a very wide range of properties which allow for their extensive use in society. The image is of Styrofoam packaging peanut. The reference source URL is.

5: Polymer Material Selection | Polymer Advisor Selector Tool | DuPont USA

"ACS Applied Polymer Materials will publish excellent work in the field of polymer materials science, with emphasis on application. It will enhance the ACS Applied Materials & Interfaces family by bringing authors the ability to publish their work in a forum that is dedicated to the field of applied polymer science."

Polymers Polymers may have the qualities of liquids or solids. Silly Putty is a great example of a polymer with both: Treats like gummies and Jell-O contain gelatin, an edible polymer. Gelatin is also used to make drug capsules and many other things, including film for movies and cameras. Rubber is a polymer. This plastic bottle and the shampoo in it contain polymers. A polymer the name means "many parts" is long chain molecule made up many repeating units, called monomers. Polymers can be natural organic or synthetic. An expert in polymer science can find work in almost any industry. How much polymer does it take to turn water into goop? Polymer chains interpenetrate each other when dissolved in solvent. A high molecular weight polymer can be 10,000 monomers long! Dan Janiak, an alumnus of the MSE graduate program, developed a polymer called a molecularly-imprinted hydrogel that could be used to filter viruses out of your blood! The black areas represent pits or low points in the film. Designed by the research group of Professor Robert M. Briber in collaboration with NIST, these polymers can be used to create patterns that would serve as templates in the manufacture of nano- or microscale electronics. Watch a materials video demonstration about polymers: When dropped, the "happy" ball will bounce while the "sad" one will not. This is because the "happy" ball is made of neoprene, an elastic polymer, and the "sad" ball is made of polynorborene, a polymer material designed to absorb energy. The polynorborene ball absorbs the impact when it hits a surface, causing it to "drop like a stone. They are generally used in the form of small particles that are crosslinked so they will form gel rather than completely dissolving. The polymer gel absorbs water and the charges along the chain repel each other, stretching out the chain and enhancing the swelling of the gel. Super absorbent polymers can readily absorb times their volume in water! Super absorbant polymers are used in products like disposable diapers, for cleaning up water based environmental spills, and for preventing rain water runoff in agricultural areas. This is a relatively low molecular weight polymer and many applications for polyethylene would require a significantly higher molecular weight to attain good mechanical properties. Learn how plastics are used in vehicles, electronics, packaging, and construction; how plastics contribute to safety and can help you save energy; how they are recycled; and how people are doing more with less plastic than before.

6: Polymer materials - EPSRC website

In the Polymer Materials area, there is a good 'people balance' featuring established leaders, fellows and Centres for Doctoral Training (CDTs). We will look to maintain this and share best practice. Other aims include maintaining access to large facilities and encouraging links to users to accelerate impact.

As well as directly helping to meet a range of real-world challenges, the area enables a number of other research areas. In this Delivery Plan period, the focus will be to build on core strengths and drive links with healthcare challenges and opportunities arising from advanced materials research. Specifically, we aim to: This will include links to the Sir Henry Royce Institute, to ensure that we understand government and industrial drivers for communicating important technical challenges and share appropriate opportunities. Ensure the community is fully engaged in EPSRC research opportunities in areas where there is strong industrial interest in polymers e. We will look to maintain this and share best practice. Other aims include maintaining access to large facilities and encouraging links to users to accelerate impact. Evidence source 4 It also demonstrated that polymer chemistry and polymer-based materials science in the UK are currently vibrant and have a strong international reputation. Synthesis and characterisation, in particular, remain important. This area is well-balanced demographically with a good range of fellows, first grants and leading established academics. Supporting this, a polymer-related CDT has been funded at the University of Sheffield as well as one in composite materials at the University of Bristol. These will help ensure the future supply of trained people. The development of polymers for healthcare applications e. About a quarter of the total value of the Polymer Materials portfolio is directly relevant to health. In terms of project partners on grants, the top contributors are in the personal care, chemicals and pharma sectors. Advanced materials has been highlighted as one of 12 potentially economically disruptive technologies Evidence source 7 , with medicine highlighted as one of the most promising areas for adoption of advanced nanomaterials; new Polymer Materials could contribute to this. This area has long-term links to all Outcomes and Ambitions, with immediate impact in the Productive Nation Outcome. Introduce the next generation of innovative and disruptive technologies P5: Transform to a sustainable society, with a focus on the circular economy R4: Manage resources efficiently and sustainably We will continue to see steady growth in the sub-topic of renewable and biodegradable Polymer Materials. Manufacturing and engineering could provide even more opportunities to deliver impact. Improve prevention and public health H4: Develop future therapeutic technologies Links to the life sciences will lead to fundamental advances in the development of Polymer Materials for tissue regeneration, drug delivery and medical implants.

7: Polymer Materials

Polymer Materials: 3D Printing Opportunities - is the most comprehensive analysis of polymer 3D printing technology available in the world, with focus on evolution in polymer print materials and subsequent market opportunities associated with them.

8: Polymer Project

Polymer is an interdisciplinary journal dedicated to publishing innovative and significant advances in Polymer Physics, Chemistry and Technology. We welcome submissions on polymer hybrids, nanocomposites, characterisation and self-assembly.

9: Polymer engineering - Wikipedia

Polymeric Materials A polymer is a large molecule, or macromolecule, composed of millions of repeated linked units, each a relatively light and simple molecule. Because of their broad range of properties, both synthetic and natural

polymers play an essential and ubiquitous role in everyday life.

Climbing out of the pit of life Marsha Is Only a Flower (Step-Into-Reading, Step 3) Cottle, T. J. Voices in the yard. Handbook of Intelligence Studies V. 2. The case of Africa Western civilization: the modern period Din 72551 part 6 Neil gaiman terry pratchett good omens Rokujouma no shinryakusha volume 23 Somatization disorder (hysteria) Chris Long, Philadelphia Phillies The post-World War II economy Daughter of the lioness Generational rhetoric and American avant-gardism Macroeconomics, Study Guide, the World Is Flat Aplia Activation Card The triple demism of Sun Yat-sen. Occupational health and safety practices Teaching techniques and insights for instrumental music educators Babok guide Milk Producers Prompt Payment Act of 1984 Aquatic monitoring in the vicinity of the South Bay Mine, northwest Ontario Ethnic embroidery Automotive Pollution Control Public health act zimbabwe Basic general knowledge mcqs with answers The city and the battlefield: Coriolanus Parkinson disease research paper Management teams belbin 1981 Marketization models : how much buying and selling in government? Strategic peacebuilding : an overview John Paul Lederach and R. Scott Appleby Memoir of Mary L. Ware Arms and equipment guide 2e The setting of the prologue of Sophocles Antigone John Porter Letter to the President of the U. States of America A survey of metaphysics lowe The carpet people The Three Keys to Self-Empowerment Stereo sound for television A Ligurian Kitchen Bounty on a Lawman (Bounty Hunter)